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enrollments. b. The answer to a problem or the explanation for...

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7.7 Solution Concentration. In chemistry, concentration is defined as the abundance of a constituent divided by the total volume of a mixture. All of us have a qualitative idea of what is meant by concentration. Anyone who has made instant coffee or lemonade knows that too much powder gives a strongly flavored, highly concentrated drink, whereas too little results in a dilute solution that may ...

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Rhodamine B solution, dissolve 0.1 g of the solid reagent in 100 ml of 3.5 M hydrochloric acid. Ion exchange column, remove the fine material from 50-100 mesh Deacidite FF or similar resin by decantation in water. Digest twice with 2 M hydrochloric acid, wash well with water and transfer to a column 6 mm diameter with a bed depth of about 75 mm. Standard thallium stock solution, dissolve 0.065 ...

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Gelatin production from alternative non-mammalian species had grown in

importance, largely as a way to valorise by-products from fish and poultry industrial processes. Regarding fish origin, tropical and sub-tropical warm-water species might have similar rheological properties and thermostability to that of mammal gelatins; therefore, they could be used for similar applications. Recently, an ...

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For practical purposes, we can consider such solutions to be a source of "free electrons" which may be used as powerful reducing agents. In the second case, ferric salts catalyze the reaction of sodium with ammonia, liberating hydrogen and forming the colorless salt sodium amide (second equation). This is analogous to the reaction of sodium with water to give sodium hydroxide, but since ...

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Required practical 5: You should have investigated how changes in concentration affect the rates of reactions by a method involving measuring the volume of a gas produced and a method involving a change in colour or turbidity ('cloudiness!').
Particle model of the collision theory of chemical reaction rate factors
Revision Notes. Effect of changing reactant concentration in solution
Revision ...

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